

Silicon dioxide coatings from dielectric barrier discharge in a two step process

Sebastian Dahle^{1,2}, Henning Munkert¹, Wolfgang Viöl², and Wolfgang Maus-Friedrichs^{1,3}

¹ Institut für Energieforschung und Physikalische Technologien, TU Clausthal, Leibnizstraße 4, 38678 Clausthal-Zellerfeld, Germany ² Hochschule für Angewandte Wissenschaft und Kunst, Von-Ossietzky-Straße 99, 37085 Göttingen, Germany ³ Clausthaler Zentrum für Materialtechnik, TU Clausthal, Leibnizstraße 4, 38678 Clausthal-Zellerfeld, Germany



P 18.7

1. Introduction

The coating of the model substrate titanium dioxide with silicon dioxide has been investigated by means of X-ray Photoelectron Spectroscopy (XPS), Metastable Induced Electron Spectroscopy (MIES), Ultraviolet Photoelectron Spectroscopy (UPS) and Atomic Force Microscopy (AFM). The silicon was deposited using the gaseous precursor silane, which was diluted in 98.5% nitrogen for safer handling. The precipitation of the precursors before deposition through oxidation reactions was avoided by cutting the deposition process into two steps. In the first step, silicon was deposited in the form of silicon nitride. In the second step, the film was transformed into stoichiometric silicon dioxide by a second plasma treatment.

3. 200 mbar Silane/Nitrogen-Plasma

2. Experimental

For probing the surface of titanium dioxide we applied MIES and UPS using a hemispherical analyzer (VSW HA100) combined with a source for metastable helium atoms (mainly $He^{23}S_{1}$) and ultraviolet photons (HeI). Additional information for chemical analysis was obtained by XPS utilizing a commercial non-monochromatic X-ray source (Specs RQ20/38C).

The topography and roughness of the TiO_2 crystal, as well as the plasma treated samples were determined by AFM using a Veeco Dimension 3100 SPM. All measurements were performed in Tapping Mode with Al-coated silicon cantilevers (MicroMasch NSC15). The depiction, as well as the calculation of the root mean square roughness (RMS) was carried out with SPIP (Image Metrology A/S) according to ISO 4287/1.

The TiO₂ crystal was cleaned by Ar⁺ ion sputtering (60 min.) and heating (750 °C, 30 min.). Afterwards the plasma treatment, a dielectric barrier discharge was performed in two steps. The process gas in the first step was a silane gas mixture (Linde Gas) consisting of 1.5% SiH₄ (99.999%) and 98.5% N₂ (99.9996%). The process gases in the second step were O₂ (Linde Gas, 99.995%) and atmospheric air.



4. 1000 mbar Silane/Nitrogen-Plasma





1000 mbar silane plasma treatment: formation of Si₃N-particles

Avg. size of particles (w x l x h):	137 x 28 x 117 nm 352 x 195 x 213 nm, resp.
Avg. roughness:	9.53 nm
Rms roughness:	15.1 nm





6. 200 mbar Oxygen-Plasma

0,1 -

0,0



1.) 1000 mbar silane plasma treatment: formation of Si₃N-particles

2.) 200 mbar oxygen plasma treatment: transformation of Si₃N-particles, roughening of the surface

Avg. size of particles (w x l x h):	665 x 704 x 206 nm
Avg. roughness:	16.4 nm
Rms roughness:	22.2 nm

1.) Deposition step

Fragmentation ┥

Agglomeration

Si:N:O:H

 SiH_4 deluted in N_2

e.g. SiH_{4-x}+ xH

Deposition

Si:N:O:H

Pressure ependance

Stoichiometry	TiO ₂ (100)	SiH ₄ ,N ₂ -Plasma 1000 mbar 2s	O ₂ -Plasma 200 mbar 10s
Ti	25.5 %	9.2 %	6.8 %
0	74.5 %	30.6 %	65.7 %
С	0.0 %	12.2 %	2.4 %
Ν	0.0 %	12.7 %	2.2 %
Si	0.0 %	35.4 %	23.0 %



1000 mbar silane plasma treatment: formation of Si₃N-particles 1000 mbar air plasma treatment: agglomeration and transformation of Si₃N-particles

Avg. size of particles (w x I x h):626 x 891 x 156 nmAvg. roughness:11.0 nmRms roughness:15.9 nm

7. 1000 mbar Oxygen-Plasma





0 / Ti	2.92	3.33	9.69	
N / Ti	-	0.36	0.09	
Layer	0.00 nm	3.22 nm	3.43 nm	

2.) Transformation step

Oxidation

TiO,

O₂ in air

Oxidation

TiO₂

SiO₂

SiO

20 15 10 5 0 -5 binding energy / eV

- 1.) 1000 mbar silane plasma treatment: formation of Si_3N -particles
- 2.) 1000 mbar oxygen plasma treatment: keeping Si₃N-particles at similar size

Avg. size of particles (w x l x h):	176 x 215 x 52 nm
Avg. roughness:	9.70 nm
Rms roughness:	12.7 nm

Stoichiometry	TiO ₂ (100)	SiH ₄ ,N ₂ -Plasma 1000 mbar 2s	O ₂ -Plasma 1000 mbar 2s
Ti	33.3 %	21.3 %	20.3 %
0	66.7 %	53.3 %	60.5 %
С	0.0 %	13.4 %	8.7 %
Ν	0.0 %	2.6 %	2.0 %
Si	0.0 %	9.4 %	8.5 %
O / Ti	2.00	2.50	2.98
N / Ti	-	0.28	0.24
Layer	0.00 nm	1.01 nm	0.75 nm

9. Literature

	P.
	S.
	-

[1]

[2]

P. Van Cappellen et. al., Geochim. Cosmochim. Acta 57, 3505-3518 S. Dahle, L. Wegewitz, A. Weber, W. Maus-Friedrichs, International Conference on Plasma Surface Engineering (2012)



Generation of

 $O_x / N_2 O_x / NO_x$

TiO_{2-x}

We are thankful for the technical assistance of Aaron Arendt and Dana Schulte Genannt Berthold and for the financial support of the Deutsche Forschungsgemeinschaft (DFG) under Project No. MA 1893/18-1 and VI 359/9-1.

s.dahle@pe.tu-clausthal.de h.munkert@pe.tu-clausthal.de



8. Model