TU Clausthal **O** 59.4 Deposition of silver nanoparticles on to thin films of ionic liquid on SiO₂

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1. Motivation

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Room-temperature ionic liquids (RT-ILs) have recently attracted much attention as a new organic solvents and functionalized materials. As ionic liquids have very low vapour pressures, at room temperature they can be employed in vacuum experiments as fluid substrates or solvents. At elevated temperatures it is possible to evaporate them in UHV and deposit thin ionic liquid films. These thin films could be used now as buffer layers for nanoparticle growth. In this work the influences of monolayers of 1-ethyl-3methylimidazolium bis(trifluoromethylsulfonyl)imide [EMIm]Tf₂N on the growth of silver nanoparticles on SiO₂ were investigated. The chemical composition and electronic structure of the silver-IL-SiO₂ interface was characterized by Metastable Impact Electron Spectroscopy (MIES) and UPS (He I). The particles size distribution was determined with Atomic Force Microscopy (AFM). We compare these results with measurements on silver deposited on pure SiO_2 .

2. Experimental

XPS: X-ray source (AIK_{α}) **UPS**: He I (21.2 eV) **MIES**: He* $2^{3}S_{1}$ (19.8 eV) UHV systems with base pressure $< 2.0 \times 10^{-10}$ Torr Substrate: SiO₂/Si(100) [1-Ethyl-3-Methyl-Imidazolium]Tf₂N films are ~0.5-1 nm Ag covering between 0.1 and 1 nm

RT-IL's film preparation:

The RT-IL films were prepared by evaporating of ultrapure Ionic Liquids in UHV system at 140°C for 120 sec onto the SiO₂ substrate. Ag particles deposition:

After evaporating RT-IL onto SiO₂ Ag was evaporated on the thin film by using metal evaporator. The evaporation occurs in UHV chamber with the base pressure <2*10⁻⁸ Torr.

4. $Ag+[EMIm]Tf_2N \text{ on }SiO_2$

MIES MIES — SiO₂ $--SiO_2$ Ag $3d_{5/2}$ and $3d_{3/2}$ - Peaks ____ [EMIM]Tf₂N /SiO₂ Ag 3d_{5/2} and 3d_{3/2} Peak $---SiO_2$ +Ag XPS - Ag+[EMIM]Tf₂N /SiO₂ XPS AIK units units units arb arb. count rate / arb. units rate arb. rate count count count Im MMM Mumm 370 375 380 binding energy / eV 390 370 375 380 binding energy / eV 385 385 390 360 365 370 365 360 20 20 15 15 10 binding energy / eV binding energy / eV Si 2p - Peak Si 2p - Peak XPS UPS XPS UPS — SiO₂ AI K_{α} —— SiO₂ Al K -SiO₂ --SiO₂ —— [EMIM]Tf₂N /SiO₂ - SiO₂+Ag - [EMIM]Tf₂N / SiO₂ $---SiO_2$ +Ag - Ag+[EMIM]Tf₂N /SiO₂ -- [EMIM]Tf₂N + Ag / SiO₂ S

3. Ag on SiO_2





Ag / SiO_2

Ag + [EMIm]Tf₂N / SiO₂

typical peaks of [EMIm]Tf₂N at 11 eV and 12 eV

- peaks of IL at 11 eV and 12 eV decrease

MIES Results

- IL evaporation:
- Ag evaporation: O 2p decreases typical Auger process change from AD to AN

Ag evaporation: - O 2p decreases

-13.1 nm

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UPS Results

L evaporation:

- O 2p decreases

- O 2p decreases



IL evaporation: 30 25 count - round particles average size particle number 12 15 18 21 24 6 9

Size, nm

He

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typical peaks of [EMIM]Tf₂N at 11eV and 12eV

- additional O 2p decreases
- peaks of IL at 11eV and 12eV decrease



- 18.6 nm

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